

(+)-CORYNOLINE 11-O-SULFATE FROM *CORYDALIS INCISA**

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Key Word Index—*Corydalis incisa*; Papaveraceae; alkaloid; (+)-corynoline 11-O-sulfate; hydrobenzo[c]phenanthridine.

Abstract—A novel O-sulfated alkaloid isolated from *Corydalis incisa* was characterized as the 11-O-sulfate of (+)-corynoline.

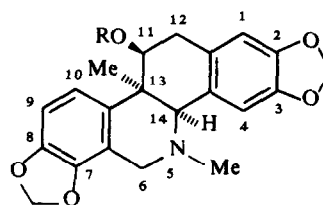
INTRODUCTION

The alkaloidal components of *Corydalis incisa* Pers (Papaveraceae) have been studied by several groups of workers [1–3] and found to be very rich in hydrobenzo[c]phenanthridine-type alkaloids. Amongst them corynoline, the main alkaloid in this plant, has been isolated as a racemate. Recent careful examination in our laboratory, however, resulted in the isolation of optically active (+)-corynoline from the same source [4]. This paper deals with the structural elucidation of an optically active corynoline homologue which was tentatively named as TN-20 [2].

RESULTS AND DISCUSSION

The tertiary non-phenolic alkaloid fraction, obtained by the previously described method [2], was chromatographed on Si gel and successive elution with hexane–EtOAc (from 4:1 to 100% EtOAc), methyl ethyl ketone and methyl ethyl ketone saturated with water. The fraction eluted with methyl ethyl ketone was further purified by chromatography on Si gel with CHCl₃–MeOH (19:1) to afford TN-20 (1). It contained a sulfur atom as indicated by combustion analysis. The UV spectrum was similar to that of (±)-corynoline (2). The MS of TN-20 (1) showed a very weak fragment peak at *m/e* 367 which corresponds to the M⁺ of (±)-corynoline (2). ((±)-Corynoline (2) showed a molecular ion peak of high intensity at *m/e* 367.) The ¹H NMR spectrum of TN-20 (1) in DMSO-*d*₆ showed a C-Me and a N-Me peak, two methylenedioxy peaks, 4 aromatic proton signals and a peak exchangeable with D₂O (Table 1, a). The ¹³C NMR spectrum in DMSO-*d*₆ contained 21 C atoms (Table 2). The multiplicities for each carbon obtained from off-resonance decoupling experiments corresponded to those for each carbon of (±)-corynoline (2) and showed 20 hydrogens attached to the carbon atoms.

The ¹H NMR and ¹³C NMR spectra of TN-20 (1) showed an alternation in the signals when the probe



- 1 R = SO₃H
2 R = H
3 R = Ac

temperature was raised to ca 100°. The ¹H NMR spectrum measured after treatment at ca 100° in DMSO-*d*₆ or DMSO-*d*₆–C₅D₅N (ca 3:1) for 10 min showed the other signals (B line) together with the signals (A line) observed at room temperature. The spectra recorded after treatment at 100° for 1–3 hr in each solution showed only the signals of the B line. These spectra were similar to those of the hydrochloride and the sulfate of (±)-corynoline (2) (Table 1, a and b). The ¹³C NMR spectrum of TN-20 recorded after accumulation for ca 2 hr at 100° in DMSO-*d*₆–C₅D₅N (ca 3:1) differed from the room temperature spectrum (Table 2). This spectrum remained unchanged as the probe temperature was lowered to room temperature (B and C line in Table 2) and it was in agreement with that of the sulfate of (±)-corynoline (2) (Table 2). The ¹H NMR spectrum recorded after treatment at 100° in C₅D₅N for 10 min differed from the room temperature spectrum and comprised only the signals of the B line (Table 1, c). Heating of TN-20 at 100° in DMSO-*d*₆–C₅D₅N (ca 3:1), and purification, yielded (+)-corynoline (2), which had an identical IR spectrum to that of naturally occurring (+)-corynoline (2). On the basis of the spectral examinations and upon the result of solvolysis, TN-20 was assumed to be the 11-O-sulfate of (+)-corynoline (2).

The 11-O-sulfate of (±)-corynoline was prepared from (±)-corynoline (2) by reaction with pyridine–sulfur trioxide complex [5] in CHCl₃. Since the synthetic sample is sparingly soluble in most organic solvents, identification of the product by IR spectrum in a solution with the natural base could not be carried out, but it was identical with the natural alkaloid by ¹H NMR (in trifluoroacetic acid) (Table 1, d) and TLC R_f value. On the

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Table 1. ^1H NMR spectral data for TN-20 (1) and corynoline (2) derivatives

		C(13) -Me	N-Me	C(12)-H ₂	C(6)-H ₂	C(14) -H	C(11) -H	OCH ₂ O	C(1) -H	C(4) -H	C(9)-H C(10)-H	NH or OH
[a in DMSO- <i>d</i> ₆]												
TN-20 (1)	A*	1.23	2.9	3.22, <i>d-d</i> (19, 4)	4.50 4.73 ABq (15)	4.5	4.73	6.09 <i>br s</i> 6.11 <i>br s</i>	6.92	7.12	7.01 7.17 ABq (8.4)	8.39
	B†	1.2	2.79	3.12, <i>d</i> (3)	4.52 4.56 ABq (16)	4.52	4.18	6.06, <i>m</i>	6.86	7.17	7.0 7.11 ABq (8)	
	C†	1.18	2.77	3.09, <i>br s</i>	4.53 ABq (16) overlapping	4.53	4.18	6.1, <i>m</i>	6.8	7.21	7.04 7.14 ABq (8)	9.67
(±)-Corynoline (2)	A*	1.06	2.19	2.95, <i>d</i> (2.8)	3.47 3.89 ABq (16)	over- lapping	3.81	6.01 <i>br s</i>	6.74	6.88	6.86 7.01 ABq (8.4)	7.16
(±)-Corynoline hydrochloride (2 HCl)	A*	1.16	2.82	3.11, <i>br s</i>	4.62 <i>br s</i>	4.79	4.19	6.09 <i>br s</i> 6.11 <i>br s</i>	6.9	7.37	7.04 7.14 ABq (8.5)	9.71
	B†	1.19	2.82	3.14, <i>br s</i>	4.62 <i>br s</i>	4.73	4.19	6.04 <i>br s</i> 6.08 <i>m</i>	6.83	7.33	6.99 7.11 ABq (8.5)	
(±)-Corynoline sulfate (2 H ₂ SO ₄)	A*	1.22	2.89	3.16, <i>br s</i>	4.66 <i>br s</i>	over- lapping	4.24	6.14 <i>br s</i>	6.93	7.36	7.02 7.18 ABq (8.5)	
	B†	1.21	2.86	3.14, <i>br s</i>	4.63 <i>br s</i>	4.5	4.22	6.07 <i>br s</i>	6.84	7.29	7.0 7.11 ABq (8)	
[b in DMSO- <i>d</i> ₆ -C ₅ D ₅ N]												
TN-20 (1)	A*	1.48	3.0	3.27, <i>d-d</i> (18, 4.5)	4.31 4.46 ABq (16)	4.43	4.94 <i>t</i> (5)	6.07 <i>m</i> 6.13 <i>br s</i>	6.77	7.23	6.98 7.4 ABq (8.4)	
	B†	1.21	2.6	3.13, <i>d</i> (3)	4.11 4.36 ABq (16)	4.14 <i>d</i> (2)	4.08	6.04 <i>m</i>	6.79	7.1	6.93 7.09 ABq (8)	
	C†	1.24	2.77	3.2, <i>d</i> (2.5)	4.39 4.56 ABq (15.5)	4.41	4.26	6.17 <i>m</i> 6.19 <i>br s</i>	6.8	7.3	7.04 7.14 ABq (8.5)	
(±)-Corynoline hydrochloride (2 HCl)	A*	1.23	2.81	3.22, <i>d</i> (3)	4.55 <i>br s</i>	4.55	4.27	6.16 <i>br s</i>	6.87	7.43	7.0 7.14 ABq (8.4)	
	B†	1.2	2.63	3.16, <i>d</i> (3)	4.26 4.34 ABq (16)	4.13 <i>d</i> (3)	4.22	6.06 <i>br s</i>	6.79	7.21	6.93 7.1 ABq (8.4)	
[c in C ₅ D ₅ N]												
TN-20 (1)	A*	1.58	3.07	3.39, <i>d-d</i> (18, 4.5)	4.51 4.56 ABq (16)	4.53	5.36 <i>t</i> (5)	5.64 <i>br s</i> 5.83 <i>br s</i>	6.53	7.38	6.84 7.52 ABq (8.4)	
	B†	1.2	2.29	3.17, <i>d-d</i> (18, 4.5)	3.66 4.14 ABq (15)	3.59	4.14	5.91 <i>m</i> 5.94 <i>br s</i>	6.71	6.92	6.87 7.07 ABq (8)	
	C†	1.22	2.41	3.21, <i>d-d</i> (18, 4)	3.9 4.27 ABq (15)	3.86	4.22	6.0 <i>m</i> 6.07 <i>br s</i>	6.78	7.16	6.93 7.09 ABq (8)	
[d in CF ₃ CO ₂ H]												
TN-20 (1)	A*	1.49	3.11 <i>d</i> (5.5)	3.39, <i>d-d</i> (20, 3) 3.78, <i>d-d</i> (20, 1)	4.39, <i>d-d</i> (15.5, 10) 4.94, <i>d-d</i> (15.5, 2)	4.33 <i>d</i> (9)	5.33 <i>d</i> (3)	6.07 <i>br s</i>	6.87	6.97	7.02 7.18 ABq (8.5)	7.78
	A*	1.53	3.14 <i>d</i> (5.5)	3.42, <i>d-d</i> (20, 4) 3.36, <i>d-d</i> (20, 1)	4.6, <i>d-d</i> (16, 10) 4.92, <i>d-d</i> (16, 2)	4.49 <i>d</i> (9)	5.69 <i>d</i> (4)	6.1 <i>br s</i>	6.84	7.01	7.03 7.11 ABq (8.5)	7.5

* Values recorded at a probe temperature of *ca* 35°.† Values recorded after the spectrum changed completely at a probe temperature of *ca* 100°.

‡ Values recorded as the probe temperature was lowered from 100° to 35°.

Values in parentheses represent approximate coupling constants (*d* = doublet, *br s* = broad singlet, ABq = AB quartet, *m* = multiplet).

basis of irradiation experiments, the proton signals of TN-20 were assigned as shown in Table 1, d. The observed couplings of H-6, H-14 and N-Me protons with the NH proton were in agreement with those obtained for (±)-corynoline acetate (3) in trifluoroacetic acid.

The 11-*O*-sulfate of (+)-corynoline was also prepared from (+)-corynoline (2) by treatment with pyridine-sulfur trioxide complex in CHCl₃ and had an identical IR spectrum to that of the natural alkaloid. TN-20 was

therefore shown to be (+)-corynoline 11-*O*-sulfate.

The ^1H NMR and ^{13}C NMR spectra were examined in order to determine the structure for TN-20 in solution. The ^1H NMR spectrum of TN-20 in DMSO-*d*₆ showed the signals at lower field relative to those observed for (±)-corynoline (2). A similar low field shift was observed between the salt and free base of (±)-corynoline (Table 1, a). In the ^{13}C NMR spectrum in DMSO-*d*₆, the signals for C-1a, C-6a and C-10a were found at

Table 2. Carbon-13 spectral data for

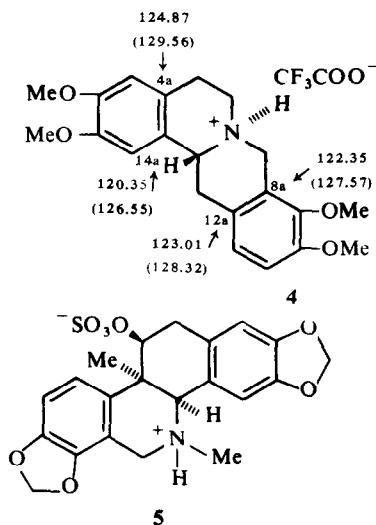
		C-Me	C-12	C-Me	N-Me	C-6	C-14	C-11	OCH ₂ O	C-1
[in DMSO- <i>d</i> ₆]										
TN-20 (1)	A*	23.14	31.3	40.09	41.63	51.57	67.65	78.86	101.47	108.52
		(<i>q</i>)	(<i>t</i>)	(<i>s</i>)	(<i>q</i>)	(<i>t</i>)	(<i>t</i>)	(<i>d</i>)	(<i>t</i>)	(<i>d</i>)
(±)-Corynoline (2)	A*	23.34	36.32	40.37	42.44	52.79	68.42	74.79	100.81	107.02
		(<i>q</i>)	(<i>t</i>)	(<i>st</i>)	(<i>q</i>)	(<i>t</i>)	(<i>d</i>)	(<i>d</i>)	(<i>t</i>)	(<i>d</i>)
[in DMSO- <i>d</i> ₆ -C ₅ D ₅ N]										
TN-20 (1)	A*	24.78	31.94	40.98	42.4	50.17	67.77	78.86	101.49	108.12
	B†	22.95	35.4	40.89	41.53	52.15	68.68	74.37	101.46‡	108.32‡
	C‡	22.7	35.02	40.48	41.31	51.81	67.77	73.93	101.54	108.52
(±)-Corynoline sulfate (2 H ₂ SO ₄)	A*	22.59	34.8	40.42	41.16	51.56	67.68	73.96	101.59	108.82

The indication in parentheses shows multiplicities obtained from ^1H off-resonance experiments.

* Values recorded at a probe temperature of *ca* 35°.† Values recorded after the spectrum changed completely at a probe temperature of *ca* 100°.

‡ These values were read normally.

§ Values recorded as the probe temperature was lowered from 100° to 35°.



higher fields than those in (\pm)-corynoline (Table 2). These high field shifts at C-1a (γ position for N atom), C-10a (γ position) and C-6a (β position) might be due to the effects of *N*-protonation on ^{13}C chemical shift as observed for C-4a (γ position for N atom), C-8a (β position), C-12a (γ position) and C-14a (β position) in the ^{13}C NMR spectrum (4) of the salt of tetrahydropalmatine in CDCl_3 and trifluoroacetic acid [5]. Therefore TN-20 appears to adopt an inner salt structure in DMSO (5).

Thus, the 11-*O*-sulfate of (+)-corynoline was isolated from *Corydalis incisa* Pers. The occurrence of naturally occurring sulfates of steroid hormones and lipids from animal tissues and metabolites is well known [7]. This is the first report to our knowledge of a sulfate of an alkaloid occurring in higher plants. The fact that optically active corynoline was isolated as a sulfate is of interest in connection with the metabolism of this alkaloid in the plant.

EXPERIMENTAL

Mps are uncorr. TLC and PLC were carried out on Si gel PF₂₅₄. ^1H NMR spectra were recorded at 90 MHz at probe temps. of ca 35°, and/or ca 50° and ca 100°. Samples were dissolved in $\text{DMSO}-d_6$ or $\text{C}_5\text{D}_5\text{N}$, or $\text{DMSO}-d_6$ - $\text{C}_5\text{D}_5\text{N}$ (ca 3:1) or $\text{CF}_3\text{CO}_2\text{H}$ containing TMS as internal standard. The ^{13}C NMR spectra were measured at 22.6 MHz in 8 mm tubes at

probe temps. of ca 35°, and/or ca 50° and ca 100°. Samples were dissolved in $\text{DMSO}-d_6$ or $\text{DMSO}-d_6$ - $\text{C}_5\text{D}_5\text{N}$ (ca 3:1) containing TMS as an internal standard at a concn of ca 0.2–0.5 mol/l. Conditions of the FT NMR measurements were: spectral width, 5000 Hz; pulse width, 25–30 μsec ; acquisition time, 0.8 sec; number of data points, 8192.

Isolation and identification of TN-20 (1). The tertiary non-phenolic fraction (TN-A fraction), separated by the procedure previously described [2], was applied to a Si gel column and eluted successively with hexane-EtOAc (from 4:1 to 100% EtOAc), MeCOEt and MeCOEt satd with H_2O . Rechromatography of the fraction eluted with MeCOEt using Si gel with CHCl_3 -MeOH (19:1) afforded TN-20, colorless prisms, mp 253–254° (CHCl_3 -MeOH), $[\alpha]_D^{23} + 67^\circ$ (MeOH, *c* 0.19); IR $\nu_{\text{max}}^{\text{nujol}}$ cm^{-1} : 3100; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 237 (ϵ 4.13) and 289 (ϵ 4.02); MS *m/e*: 367 ($\text{M}^+ - \text{SO}_3$, weak), 365 (weak), 363 (weak), 349, 334, and 318. (Found: C, 56.17; H, 4.73; N, 2.98; S, 7.15. Calc. for $\text{C}_{21}\text{H}_{21}\text{NO}_8\text{S}$: C, 56.37; H, 4.73; N, 3.13; S, 7.17%).

Conversion of TN-20 (1) to (+)-corynoline (2). TN-20 (100 mg) was heated in 0.5 ml $\text{DMSO}-d_6$ - $\text{C}_5\text{D}_5\text{N}$ (ca 3:1) at probe temp. for ca 2 hr. The soln was concd under red. pres. and the residue separated by PLC on Si gel using MeOH-Et₂O (1:1) followed by C_6H_6 -Et₂O (3:7). This afforded an alkaloid which was recrystallized from MeOH-Et₂O to give (+)-corynoline (2), 57 mg, mp 180–181°, $[\alpha]_D^{23} + 132^\circ$ (CHCl_3 , *c* 2.64) identical with naturally occurring (+)-corynoline by IR (KBr).

Preparation of 11-*O*-sulfate of (\pm)-corynoline (1). (\pm)-Corynoline (0.2 g) in CHCl_3 (5 ml) was shaken at room temp. with Py-sulfur trioxide (1 g) for 2.5 hr. To this soln, Py-sulfur trioxide (1 g) was added and the mixture shaken for 2 hr and then heated under reflux for 5 hr. After cooling, the white crystalline ppt. formed was filtered off, washed with hot MeOH repeatedly to remove unreacted Py-sulfur trioxide, and dried in a vacuum desiccator for several hr. These crystals, mp 268–272°, were sparingly soluble in MeOH, CHCl_3 , *n*-BuOH, *sec*-BuOH, *n*-PrOH, and *iso*-PrOH; IR $\nu_{\text{max}}^{\text{nujol}}$ cm^{-1} : 3105; ^1H NMR ($\text{CF}_3\text{CO}_2\text{H}$): δ 1.5 (3H, s, C-Me), 3.11 (3H, d, *J* = 5.3 Hz, N-Me), 6.08 (4H, br s, $\text{OCH}_2\text{O} \times 2$), 6.87 (1H, s, Ar-H), 6.97 (1H, s, Ar-H), 7.02, 7.17 (each 1H, d, *J* = 8.4 Hz, Ar-H); MS *m/e*: 367 (weak), 365 (weak), 363 (weak), 349, 334, 318. (Found: C, 56.65; H, 4.85; N, 2.19. Calc. for $\text{C}_{21}\text{H}_{21}\text{NO}_8\text{S}$: C, 56.37; H, 4.73; N, 3.13%).

Preparation of 11-*O*-sulfate of (+)-corynoline (1). (+)-Corynoline (45 mg) in CHCl_3 (4 ml) was shaken with Py-sulfur trioxide (1 g) for 3.5 hr at room temp. and heated under reflux for 1.5 hr. After addition of Py-sulfur trioxide (500 mg), the mixture was heated under reflux for 1.5 hr. The ppt. formed was filtered off and extracted with CHCl_3 . From the CHCl_3 extract, the main product and unreacted (+)-corynoline were separated by PLC

TN-20 (1) and corynoline (2) derivatives

C-9	C-4	C-6a	C-10	C-1a	C-4a	C-10a	C-7	C-2, 3, and 8		
109.05 (d)	112.77 (d)	110.69 (s)	120.44 (d)	117.94 (s)	128.97 (s)	131.57 (s)	142.11 (s)	145.23 (s)	145.47 (s)	149.11 (s)
108.8 (d)	112.38 (d)	116.72 (s)	118.78 (d)	125.43 (s)	127.48 (s)	136.08 (s)	142.12 (s)	144.31 (s)	144.84 (s)	147.33 (s)
109.03	111.55	112.29	120.99	120.99	129.1	132.15	142.52	145.18	145.84	148.57
109.29†	113.1†	113.54†	119.47†	121.24†	128.76†	134.51†	142.21†	145.62†	146.02†	149.12†
109.25	113.33	112.63	119.92	119.41	128.52	133.92	142.64	145.35	145.63	148.96
109.34	113.53	111.84	119.53	118.99	128.81	133.47	142.68	145.65	145.65	149.16

on Si gel (C_6H_6 - Et_2O , 1:4). The main product was further purified by repeated PLC on Si gel ($CHCl_3$ - $MeOH$, 7:3). The ppt. from the reaction mixture was dissolved in $MeOH$ and then purified by PLC on Si gel ($CHCl_3$ - $MeOH$, 9:1) to afford an alkaloid identical with the above main product. The alkaloid obtained, which was recrystallized from $MeOH$ to give white crystals, mp $252-255^\circ$, was identical with natural TN-20 by IR (KBr) and MS.

REFERENCES

1. Tani, C. and Takao, N. (1962) *Yakugaku Zasshi* **82**, 594 and 598.
2. Nonaka, G., Okabe, M., Nishioka, I. and Takao, N. (1973) *Yakugaku Zasshi* **93**, 87.
3. Kametani, T., Ibara, M. and Honda, T. (1971) *Phytochemistry* **10**, 1881.
4. Takao, N., Kamigauchi, M. and Iwasa, K. (1979) *Tetrahedron* (in press).
5. Baumgarten, P. (1926) *Chem. Ber.* **59**, 1166.
6. Takao, N., Iwasa, K., Kamigauchi, M. and Sugiura, M. (1977) *Chem. Pharm. Bull. (Tokyo)* **25**, 1426.
7. Hadd, H. E. and Blickenstaff, R. T. (1969) *Conjugates of Steroid Hormones*, p. 25. Academic Press, New York and London.